

Potent Anti-HIV and Anti-HBV Activities of (-)-L- β -Dioxolane-C and (+)-L- β -Dioxolane-T and Their Asymmetric Syntheses.

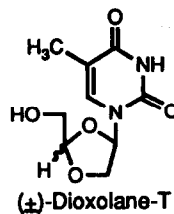
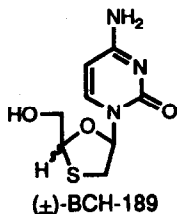
Hea O. Kim,[†] Kirupathevy Shanmuganathan,[†] Antonio J. Alves,[†] Lak S. Jeong,[†] J. Warren Beach,[†] Raymond F. Schinazi,[‡] Chien-Neng Chang,[#] Yung-Chi Cheng,[#] and Chung K. Chu^{*}

[†]Department of Medicinal Chemistry, College of Pharmacy, The University of Georgia, Athens, GA 30602, [‡]Department of Pediatrics, Emory University School of Medicine/VA Medical Center, Decatur, GA 30033 and [#]Department of Pharmacology, Yale University School of Medicine, New Haven, CT 06510.

Key Words: Anti-HIV; Anti-HBV; (+)-L- β -Dioxolane-T; (-)-L- β -Dioxolane-C; Asymmetric Synthesis

Abstract: The asymmetric syntheses of (+)-L- β -dioxolane-T and (-)-L- β -dioxolane-C were accomplished starting from 1,6-anhydro-L- β -gulopyranose, and their anti-HIV and anti-HBV activities were evaluated in human PBM cells, CEM cells and 2.2.15 cells, respectively.

(\pm)-2',3'-Dideoxy-3'-thiacytidine (BCH-189) was found to be a potent anti-HIV agent *in vitro* and its (-)-isomer [3TC, (-)-BCH-189] is undergoing clinical trials in patients with AIDS and AIDS-related complex.¹ Recently, racemic BCH-189 was also found to be a potent antiviral agent against human hepatitis B virus (HBV).² Since BCH-189 had been reported as a racemic mixture, it was of interest to synthesize the enantiomerically pure isomers. Thus, we have reported the synthesis of (+)-D-BCH-189 from D-mannose.³ Recently, we also reported the more efficient enantiomeric synthesis of (+)-D-BCH-189 from D-galactose.⁴ Interestingly, we discovered that the enantiomerically pure (+)-D-BCH-189 [EC₅₀ = 0.21-0.31 μ M in human peripheral blood mononuclear (PBM) cells infected with HIV-1] was less potent than the (\pm)-BCH-189 (EC₅₀ = 0.02-0.06 μ M).³ Thus, we synthesized the antipod, (-)-L-BCH-189 from L-gulose,⁵ and demonstrated to be more potent (EC₅₀ = 0.0018 μ M) than either the (+)-D-BCH-189 or (\pm)-BCH-189. It was also discovered that the trend for anti-HBV activity was similar to that for anti-HIV activity.⁶



(\pm)-Dioxolane-T has been reported as a moderately active anti-HIV agent (EC₅₀ = 20 μ M) in ATH8

cells⁷; however, it showed a superior activity in human PBM cells ($EC_{50} = 0.09 \mu\text{M}$).⁸ Thus, we have synthesized the enantiomerically pure (-)-D-dioxolane-T ($EC_{50} = 0.39 \mu\text{M}$) from D-mannose and found that it too was less potent than the racemate ($EC_{50} = 0.09 \mu\text{M}$) as in the case of BCH-189.⁹ A number of pyrimidine dioxolane derivatives have been synthesized and structure-activity correlation in this series showed that (+)-D-β-dioxolane-C ($EC_{50} = 0.016 \mu\text{M}$) was the most potent nucleoside against HIV-1.⁸ Now we wish to report the synthesis of the optical antipode, (-)-L-β-dioxolane-C and substantiate its potent antiviral activity against HIV and HBV.

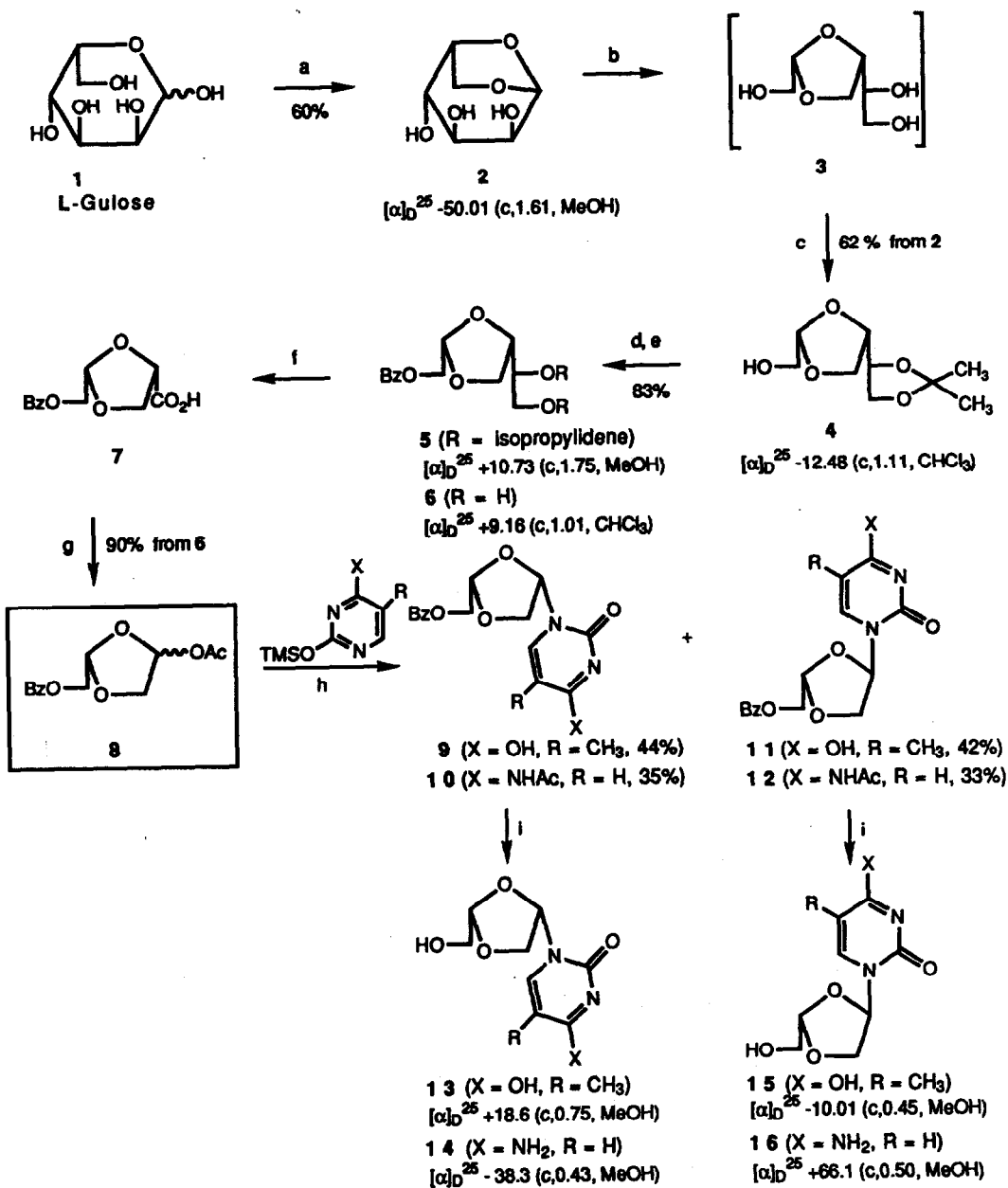
The general synthetic strategy that we have developed for the BCH-189 and dioxolane-T was used to synthesize the L-isomers 13-16 (Scheme 1). 1,6-Anhydro-L-gulose was prepared in one step from L-gulose by the treatment of L-gulose with 0.5 N HCl in 60 % yield.¹⁰ Without selective protection,^{4,5} as previously reported,⁹ 2 was directly converted to dioxolane triol 3 by NaIO_4 followed by reduction with NaBH_4 , which without isolation, was converted to isopropylidene derivative 4. Benzoylation to 5, deprotection to 6, and oxidation of diol 6 gave the acid 7. Oxidative decarboxylation of 7 with $\text{Pb}(\text{OAc})_4$ in dry THF gave the key intermediate acetate 8 in good yield. The acetate was condensed with the desired pyrimidines (silylated thymine and N-acetylcytosine) in the presence of TMSOTf to afford a mixture of anomers, which was separated by a silica gel column chromatography to give the individual isomers 9-12. Debenzoylation with methanolic ammonia gave the desired thymine and cytosine derivatives 13 and 14, respectively.¹¹

Anti-HIV and anti-HBV activities as well as cytotoxicities of 13 and 14 were evaluated in human PBM cells, CEM cells and 2.2.15 cells, respectively (Table 1).^{5,8}

Table 1. Antiviral activities and cytotoxicities.

Compounds	Anti-HIV activity in PBM cells EC_{50} (μM)	Anti-HIV activity in CEM cells EC_{50} (μM)	Anti-HBV activity in 2.2.15 cells $HBID_{50}$ (μM)	Cytotoxicity in PBM cells IC_{50} (μM)	Cytotoxicity in CEM cells IC_{50} (μM)	Cytotoxicity in Vero cells IC_{50} (μM)
(-)-L-β-dioxolane-cytosine 14	0.002	0.005	0.0005	> 10	0.26	0.10
(+)-D-β-dioxolane-cytosine	0.016 ⁸	0.009	0.01	62.0	12.3	8.3
(+)-L-β-dioxolane-thymine 13	4.81	ND	> 5	> 100	ND	> 100
(-)-D-β-dioxolane-thymine	0.39 ⁸	ND	> 5	> 100	ND	> 100
AZT	0.002	0.025	ND	> 100	14.3	28.0

As seen in Table 1, (-)-L-β-dioxolane-cytosine (14) was more potent than (+)-D-β-dioxolane-cytosine while the racemic dioxolane-thymine was more potent than either (+)-L-β-dioxolane-thymine (13) or (-)-D-β-dioxolane-thymine probably due to the additive effects of the (+)- and (-)-enantiomers. It is noted that this result is the mean of the three different assays. The cause of this discrepancy is under investigation in our laboratories.

Scheme 1^a

In summary, the asymmetric synthesis of (+)-L- β -dioxolane-T and (-)-L- β -dioxolane-C was accomplished via 8 steps from a chiral template 2. A complete evaluation of a structure-activity relationship in 1,3-dioxolane-L-nucleosides as anti-HIV and anti-HBV agents are in progress in our laboratories.

Acknowledgment. This research was supported by the U.S. Public Health Service Research grants (AI 25899, AI 32351, CA 44358, and CA 52020) and the Department of Veterans Affairs. The confirmatory anti-HIV activities in CEM cells were generously provided by Dr. Mohamed Nasr of the National Institute of Allergy and Infective Diseases.

References cited

1. a) Belleau, B.; Dixit, D.; Nguyen-Ga, N.; Kraus, J. L. *V International Conference on AIDS*, Montreal, Canada, June 4-9, 1989, paper No. T.C.O. 1. b) Coates, J. A. V.; Cammack, N.; Jenkinson, H. J.; Mutton, I. M.; Pearson, B. A.; Storer, R.; Cameron, J. M.; Penn, C. R. *Antimicrob. Agents Chemother.* 1992, 36, 202. c) Schinazi, R. F.; Chu, C. K.; Peck, A.; McMillan, A.; Mathis, R.; Cannon, D.; Jeong, L. S.; Beach, J. W.; Choi, W.-B.; Yeola, S.; Liotta, D. C. *Antimicrob. Agents Chemother.* 1992, 36, 672.
2. Doong, S.-L.; Tsai, C.-H.; Schinazi, R. F.; Liotta, D. C.; Cheng, Y.-C. *Proc. Natl. Acad. Sci. U.S.A.* 1991, 88, 8495.
3. Chu, C. K.; Beach, J. W.; Jeong, L. S.; Choi, B. G.; Comer, F. I.; Alves, A. J.; Schinazi, R. F. *J. Org. Chem.* 1991, 56, 6504.
4. Jeong, L. S.; Alves, A. J.; Carrigan, S. W.; Kim, H. O.; Beach, J. W.; Chu, C. K. *Tetrahedron Lett.* 1992, 33, 595.
5. Beach, J. W.; Jeong, L. S.; Alves, A. J.; Pohl, D.; Kim, H. O.; Chang, C.-N.; Doong, S.-L.; Schinazi, R. F.; Cheng, Y.-C.; Chu, C. K. *J. Org. Chem.* 1992, 57, 2217.
6. Chang, C.-N.; Doong, S.-L.; Zhou, J. H.; Beach, J. W.; Jeong, L. S.; Chu, C. K.; Tsai, C. H.; Cheng, Y. C. *J. Biol. Chem.* in press.
7. Norbeck, D. W.; Spanton, S.; Broder, S.; Mitsuya, H. *Tetrahedron Lett.* 1989, 30, 6263.
8. Kim, H. O.; Ahn, S. K.; Alves, A. J.; Beach, J. W.; Jeong, L. S.; Choi, B. G.; Van Roey, P.; Schinazi, R. F.; Chu, C. K. *J. Med. Chem.* 1992, 35, 1987.
9. Chu, C. K.; Ahn, S. K.; Kim, H. O.; Beach, J. W.; Alves, A. J.; Jeong, L. S.; Islam, Q.; Van Roey, P.; Schinazi, R. F. *Tetrahedron Lett.* 1991, 32, 3791.
10. Evans, M. E.; Parrish, F. W. *Carbohydr. Res.* 1973, 28, 359.
11. All key intermediates and final compounds 13-16 in Scheme I gave correct elemental analyses ($\pm 0.4\%$). These compounds 13-16 were spectroscopically identical to those reported previously with the exception of the optical rotation which was of an opposite sign but of equal value.

(Received in USA 11 June 1992; accepted 20 August 1992)